Polarized spectroscopy of Sm³⁺ ions in monoclinic KGd(WO₄)₂ crystals

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Abstract. We report on a polarization-resolved spectroscopic study of Sm³⁺-doped monoclinic KGd(WO₄)₂ crystals. The transition probabilities for Sm³⁺ ions were calculated using a modified Judd-Ofelt theory. For the ${}^{4}G_{5/2} \rightarrow {}^{6}H_{9/2}$ transition in the red spectral range, the stimulated-emission cross-section is 5.59×10^{-21} cm² at 649.0 nm (for light polarization $E \parallel N_{p}$) and the luminescence lifetime of the ${}^{4}G_{5/2}$ state is 719 µs (0.4 at.% Sm³⁺-doping). Sm:KGd(WO₄)₂ is promising for orange and red lasers.

1 Introduction

Trivalent samarium ions (Sm3+) possess an electronic configuration of [Xe]4f⁵, with a group of lower-lying ⁶F_J and ⁶H_J multiplets (⁶H_{5/2} is the ground-state) and a metastable state ${}^{4}G_{5/2}$. This energy-level structure gives rise to multiple emissions in the visible and near-infrared, among which the transitions in orange (${}^{4}G_{5/2} \rightarrow {}^{6}H_{7/2}$), red $({}^{4}G_{5/2} \rightarrow {}^{6}H_{9/2})$ and deep-red $({}^{4}G_{5/2} \rightarrow {}^{6}H_{11/2})$ are of main interest for the development of visible laser sources [1]. So far, only a few Sm³⁺-doped crystals (e.g., LiYF₄ and SrAl₁₂O₁₉) were studied in this regard. Monoclinic double tungstates represent a family of laser host matrices being very suitable for doping with trivalent rare-earth (RE³⁺) ions. They feature high absorption and emission crosssections of RE³⁺ ions with a strong polarization anisotropy and weak concentration quenching of luminescence. Visible laser emission was recently achieved from a stoichiometric KEu(WO₄)₂ crystal [2].

In the present work, we report on a polarization-resolved spectroscopic study of Sm^{3+} ions in monoclinic potassium gadolinium double tungstate (KGd(WO₄)₂) crystal, for the first time, with the goal of developing novel gain media for visible lasers.

2 Crystal growth

A series of Sm³⁺-doped KGd(WO₄)₂ crystals was grown by the Top-Seeded Solution Growth (TSSG) method using K₂W₂O₇ as a solvent and [010]-oriented seeds. Three doping levels were studied, 0.4, 0.8 and 20 at.% of Sm³⁺. The crystals are monoclinic (sp. gr. C⁶_{2h} – C2/c). The Sm³⁺ ions replace for the host-forming Gd³⁺ cations (ionic radii: $R_{\rm Gd} = 1.053$ Å and $R_{\rm Sm} = 1.079$ Å for VIIIfold oxygen coordination) in a single type of sites (C₂). The Sm:KGd(WO₄)₂ crystal is optically biaxial, and its optical properties are described in the optical indicatrix frame $N_{\rm p}$, $N_{\rm m}$ and $N_{\rm g}$.

3 Polarized spectroscopy

The polarized absorption cross-section, σ_{abs} , spectra of Sm³⁺ ions in the violet and blue spectral ranges are shown in Fig. 1. The spectra are strongly polarized, with $\boldsymbol{E} \parallel N_{\rm m}$ being the preferable pump polarization. In the spectral range addressed by commercial 2 ω -OPSLs, the maximum σ_{abs} is 0.76×10^{-20} cm² at 487.8 nm corresponding to an absorption bandwidth (FWHM) of only 2.7 nm. This absorption peak corresponds to the spin-forbidden ($\Delta S \neq 0$) $^{6}\text{H}_{5/2} \rightarrow ^{4}\text{M}_{15/2}$ transition. At shorter wavelengths, an intense absorption band is related to the spin-allowed ($\Delta S = 0$) $^{6}\text{H}_{5/2} \rightarrow ^{6}\text{P}_{3/2}$ transition. The corresponding σ_{abs} is about an order of magnitude higher, 8.37×10^{-20} cm² at 404.5 nm (FWHM, 0.9 nm).



Fig. 1. (a,b) Polarized absorption cross-sections, σ_{abs} , for Sm³⁺ ions in the KGd(WO₄)₂ crystal in the blue-violet spectral range.

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Table 1. Probabilities of spontaneous radiative transitions for Sm^{3+} ions in the KGd(WO₄)₂ crystal calculated using the Judd-Ofelt theory: λ_{em} – mean emission wavelength, $A_{JJ'}$ – transition probability (ED – electric dipole, MD – magnetic dipole), $\beta_{JJ'}$ – luminescence branching ratios.

${}^4\mathrm{G}_{5/2} \rightarrow$	λ _{em} (nm)	βjj (%)	Ајј [,] (s ⁻¹)
⁶ F _{11/2}	1355.6	0.05	0.66 ^{ED}
⁶ F9/2	1156.9	0.8	10.59 ^{ED}
⁶ F _{7/2}	1026.3	1.0	12.18 ^{ED} +1.74 ^{MD}
${}^{6}F_{5/2} {+}^{6}F_{3/2} {+} \\ {}^{6}H_{15/2} {+}^{6}F_{1/2}$	903.0	8.9	$104.3^{\text{ED}} + 16.75^{\text{MD}}$
⁶ H _{13/2}	794.2	0.9	12.82 ^{ED}
⁶ H _{11/2}	716.3	8.4	114.1 ^{ED}
⁶ H _{9/2}	655.0	40.2	548.2 ^{ED}
⁶ H _{7/2}	607.2	33.8	434.5 ^{ED} +25.54 ^{MD}
⁶ H _{5/2}	564.1	6.0	49.93 ^{ED} +31.27 ^{MD}

Based on the measured polarized absorption spectra, the 4f – 4f transition probabilities for Sm³⁺ in the KGd(WO₄)₂ crystal were calculated using the modified Judd-Ofelt (mJ-O) theory accounting for configuration interaction [3]. The obtained intensity parameters are $\Omega_2 = 8.027$, $\Omega_4 = 7.210$, $\Omega_6 = 2.322$ [10⁻²⁰ cm²] and $\alpha = -0.017$ [10⁻⁴ cm]. The probabilities of spontaneous radiative transitions from the metastable ${}^4G_{5/2}$ Sm³⁺ state are listed in Table 1. The radiative lifetime, τ_{rad} , of this manifold is 734 µs. For laser transitions, ${}^4G_{5/2} \rightarrow {}^6H_{9/2}$ (in the red) and ${}^4G_{5/2} \rightarrow {}^6H_{7/2}$ (in the orange), the luminescence branching ratios $\beta_{JJ'}$ are 40.2% and 33.8%, respectively.



Fig. 2. Polarized stimulated-emission (SE) cross-sections, σ_{SE} , of Sm³⁺ ions in the KGd(WO₄)₂ crystal: (a) orange, the ${}^{4}\text{G}_{5/2} \rightarrow {}^{6}\text{H}_{7/2}$ transition; (b) red, the ${}^{4}\text{G}_{5/2} \rightarrow {}^{6}\text{H}_{9/2}$ transition.

The stimulated-emission (SE) cross-sections, σ_{SE} , for orange and red Sm³⁺ emissions were calculated using the Füchtbauer-Ladenburg equation based on the measured polarized luminescence spectra and the τ_{rad} and β_{JJ} values derived from the mJ-O theory. The σ_{SE} spectra exhibit a strong polarization anisotropy which is a prerequisite for linearly polarized laser emission, as shown in Fig. 2. For the ${}^{4}G_{5/2} \rightarrow {}^{6}H_{9/2}$ transition, the peak σ_{SE} is 5.59×10^{-21} cm² at 649.0 nm and the emission bandwidth (FWHM) is 1.4 nm (for $E \parallel N_{p}$). For the ${}^{4}G_{5/2} \rightarrow {}^{6}H_{7/2}$ transition, the SE cross-sections are lower, with a peak value of 3.95×10^{-21} cm² at 610.0 nm and an emission bandwidth of 2.1 nm (for $E \parallel N_{m}$).

The RT luminescence decay curves from the ${}^{4}G_{5/2}$ Sm³⁺ state were measured for different doping levels, see Fig. 3. For 0.4 at.% Sm³⁺, the luminescence decay is nearly single exponential in agreement with a single type of sites for the dopant ions (C₂ symmetry), and the luminescence lifetime is 719 µs in good agreement with the radiative

one. On increasing the doping level (*i.e.*, decreasing the distances between the active ions), the cross-relaxation processes $({}^{4}G_{5/2} + {}^{6}H_{5/2} \rightarrow {}^{6}F_{J} + {}^{6}F_{J'})$ between adjacent Sm³⁺ ions are enhanced leading to luminescence quenching and non-single-exponential decay.



Fig. 3. Luminescence decay curves from the ${}^{4}G_{5/2}$ Sm³⁺ state in the KGd(WO₄)₂ crystal for various doping levels, *circles* – experimental data, *curves* – fits using the Inokuti-Hirayama model.

The decay curves were fitted using the Inokuti-Hirayama model for multipolar interactions [4], yielding the best-fit parameters $\tau_0 = 750 \ \mu s$ (the intrinsic lifetime), $R_0 = 8.2 \ \text{Å}$ (the critical distance for energy transfer), and s = 8 (the parameter for dipole-quadrupole interactions). The average luminescence lifetime $<\tau_{\text{lum}}>$ decreased to 686 $\ \mu s$ for 0.8 at.% Sm³⁺ and further to 154 $\ \mu s$ for 20 at.% Sm³⁺.

4 Conclusion

To conclude, Sm^{3+} -doped KGd(WO₄)₂ crystals are appealing for the development of visible lasers due to (i) relatively broad absorption around ~480 nm, (ii) high SE cross-sections with a strong polarization-anisotropy in the red and orange spectral ranges, (iii) a relatively long luminescence lifetime of the ${}^{4}\text{G}_{5/2}$ metastable state and weak luminescence self-quenching via cross-relaxation. Further evaluation of the potential of these crystals require the study of excited-state absorption (ESA) in the visible. *Funding*. Agence Nationale de la Recherche (ANR-22-CE08-0025-01, NOVELA); Contrat de plan État-Région (CPER) de Normandie.

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